

**Assay** Not less than 98.5% and not more than 101.5% of  $C_9H_{11}NO_3$ , calculated on the dried basis.

**Lead** Not more than 5 mg/kg.

**Loss on Drying** Not more than 0.3%.

**Optical (Specific) Rotation**  $[\alpha]_D^{20}$ : Between  $-11.3^\circ$  and  $-12.3^\circ$ , calculated on the dried basis; or  $[\alpha]_D^{25}$ : Between  $-10.0^\circ$  and  $-11.0^\circ$ , calculated on the dried basis.

**Residue on Ignition** Not more than 0.1%.

## TESTS

**Assay** Transfer about 400 mg of sample, previously dried at  $105^\circ$  for 3 h and accurately weighed, into a 250-mL flask. Dissolve the sample in about 50 mL of glacial acetic acid, add 2 drops of crystal violet TS, and titrate with 0.1 N perchloric acid to a blue-green endpoint.

**Caution:** Handle perchloric acid in an appropriate fume hood.

Perform a blank determination (see *General Provisions*), and make any necessary correction. Each milliliter of 0.1 N perchloric acid is equivalent to 18.12 mg of  $C_9H_{11}NO_3$ .

**Lead** Determine as directed under *Lead Limit Test*, Appendix IIIB, using a *Sample Solution* prepared as directed for organic compounds, and 5  $\mu$ g of lead (Pb) ion in the control.

**Loss on Drying** Determine as directed under *Loss on Drying*, Appendix IIC, drying a sample at  $105^\circ$  for 3 h.

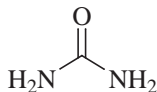
**Optical (Specific) Rotation** Determine as directed under *Optical (Specific) Rotation*, Appendix IIB, using a solution containing 5 g of a previously dried sample in sufficient 1 N hydrochloric acid to make 100 mL.

**Residue on Ignition** Determine as directed under *Residue on Ignition*, Appendix IIC, igniting a 2-g sample.

**Packaging and Storage** Store in well-closed containers.

## Urea

Carbamide



$CH_4N_2O$

Formula wt 60.06

CAS: [57-13-6]

## DESCRIPTION

Urea occurs as a colorless to white, prismatic, crystalline powder or as small, white pellets. It is commonly produced from  $CO_2$  by ammonolysis or from cyanamide by hydrolysis. It is freely soluble in water and in boiling alcohol, but practi-

cally insoluble in chloroform and in ether. It melts at a range of  $132^\circ$  to  $135^\circ$ .

**Function** Fermentation aid.

## REQUIREMENTS

### Identification

A. Heat about 500 mg of sample in a test tube until it liquefies. Ammonia vapor is produced. Continue heating until the liquid becomes turbid, and then cool. Dissolve the fused mass in a 1:10 sodium hydroxide solution:water mixture. Add 1 drop of cupric sulfate TS. A red-violet colored solution develops.

B. Dissolve 100 mg of sample in 1 mL of water, and add 1 mL of nitric acid. A white precipitate of urea nitrate forms.

**Assay** Not less than 99.0% and not more than 100.5% of  $CH_4N_2O$ .

**Alcohol-Insoluble Matter** Not more than 0.04%.

**Chloride** Not more than 0.007%.

**Lead** Not more than 5 mg/kg.

**Loss on Drying** Not more than 1.0%.

**Residue on Ignition** Not more than 0.1%.

**Sulfate** Not more than 0.01%.

## TESTS

**Assay** Transfer about 500 mg of sample, accurately weighed, into a 200-mL volumetric flask, and dissolve it in 100 mL of water, dilute to volume with water, and mix. Pipet 2 mL of this solution into a semimicro Kjeldahl digestion flask, and proceed as directed in *Method II* under *Nitrogen Determination*, Appendix IIIC. Heat the sample until it begins to fume, and then heat for 1 additional hour. Each milliliter of 0.01 N acid is equivalent to 0.3003 mg of  $CH_4N_2O$ .

**Alcohol-Insoluble Matter** Dissolve about 5 g of sample, accurately weighed, in 50 mL of warm alcohol. If any residue remains, filter the solution through a tared filter, wash the residue, and filter with 20 mL of warm alcohol. Dry at  $105^\circ$  for 1 h. Cool in a desiccator, and weigh.

**Chloride** Determine as directed in the *Chloride Limit Test* under *Chloride and Sulfate Limit Tests*, Appendix IIIB, using 0.2 g of sample, and 14  $\mu$ g of chloride (Cl) ion in the control.

**Lead** Determine as directed under *Lead Limit Test*, Appendix IIIB, using 2 g of sample, and 10  $\mu$ g of lead (Pb) ion in the control.

**Loss on Drying** Determine as directed under *Loss on Drying*, Appendix IIC, drying a sample at  $105^\circ$  for 3 h.

**Residue on Ignition** Determine as directed in *Method I* under *Residue on Ignition*, Appendix IIC, igniting a 1-g sample.

**Sulfate** Determine as directed in the *Sulfate Limit Test* under *Chloride and Sulfate Limit Tests*, Appendix IIIB, using 2 g of sample, and 200  $\mu$ g of sulfate ( $SO_4$ ) ion in the control.

**Packaging and Storage** Store in a well-closed container.